

# Development and characterization of a toluene permeation tube in a dynamic gas generation system for calibration of an online gas chromatography at ppb level

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## INTRODUCTION

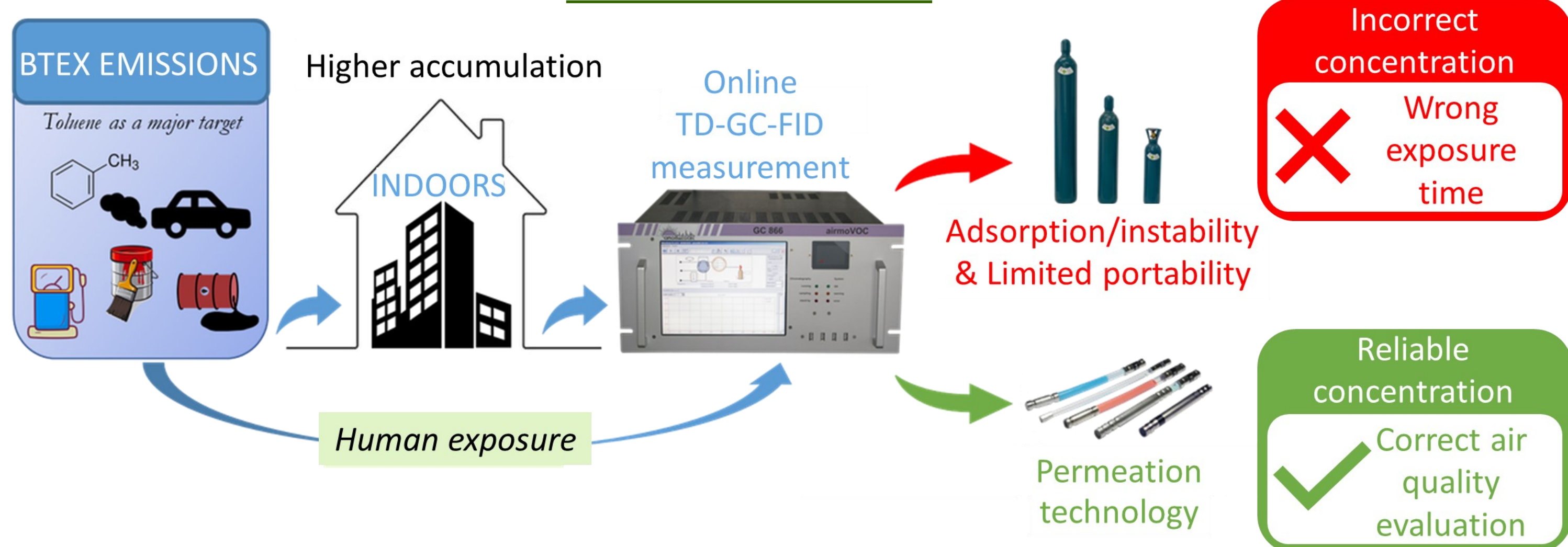


Figure 1: Improving indoor BTEX exposure assessment through reliable quantification using permeation technology and online TD-GC-FID.

- BTEX (benzene, toluene, ethylbenzene and xylenes) contribute to ozone and secondary organic aerosol formation [1]. Indoor toluene concentration (few to tens of ppb) often exceeds outdoor levels (sub to few of ppb) [2]. Gaseous pressurized cylinders suffer from adsorption, instability, and limited portability, while permeation technology (dynamic standard) can be efficient for in-situ calibration at trace levels (Figure 1).
- Advantages of using real-time GC-based methods for analysis of emissions from permeation devices outweigh those of the weighing method (Figure 2).

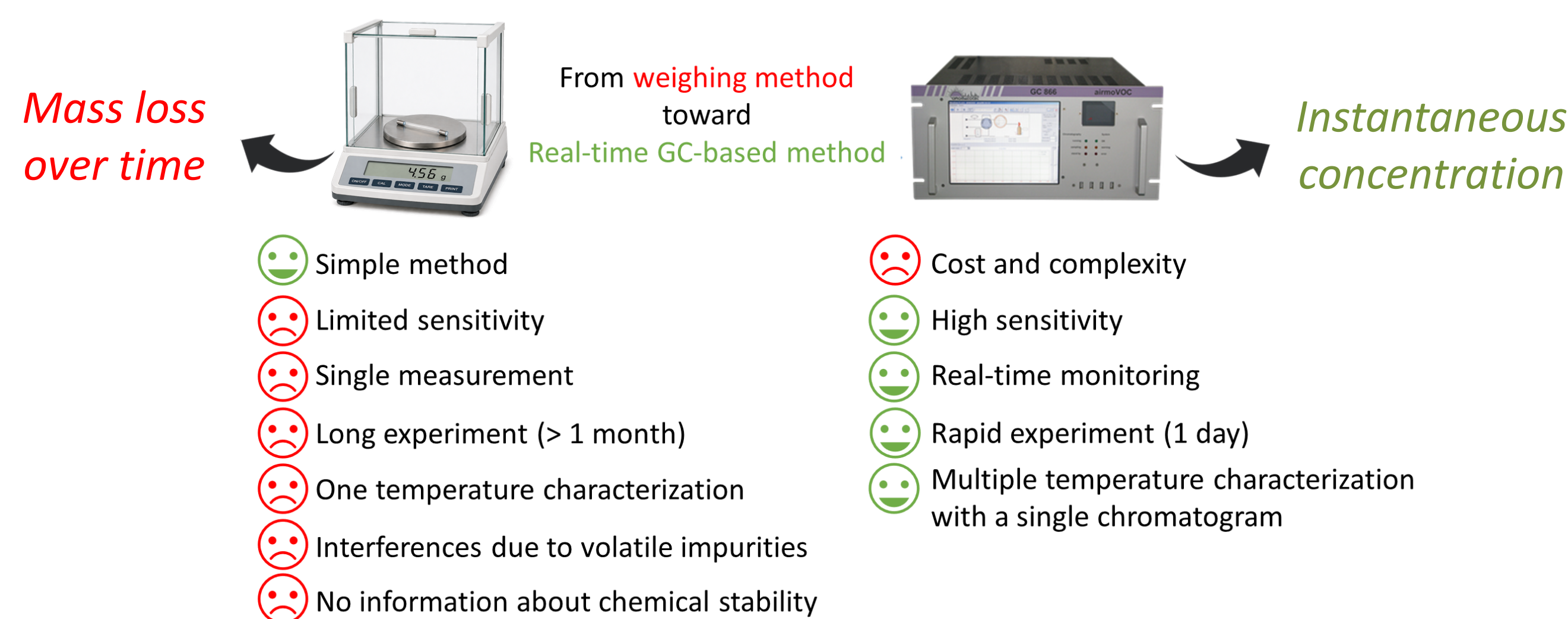
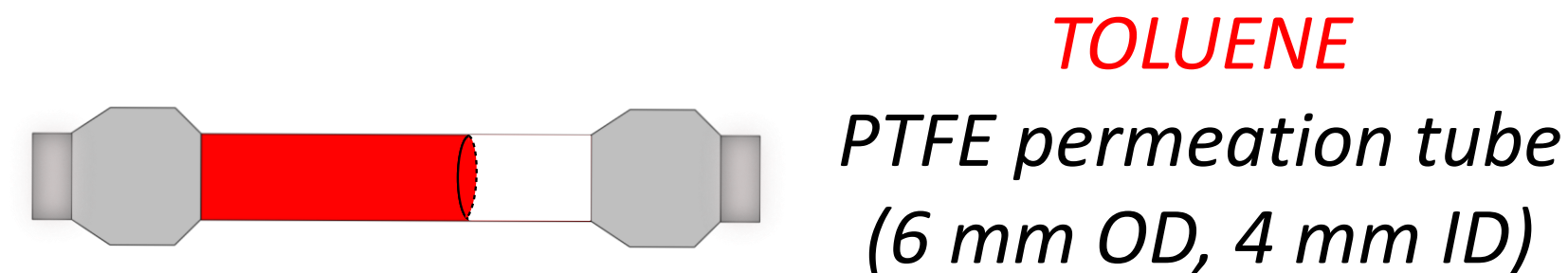


Figure 2: Advantages and drawbacks of weighing versus real-time GC methods for permeation characterization.

- Trace quantification using analytical tools like Thermodesorption-Gas Chromatography-Flame Ionization Detector (TD-GC-FID) coupled to a gas generator depends on calibration quality [3]. A toluene permeation tube was designed together with a gas generator that contains it and produces toluene at low-ppb concentrations using suction approach.

## METHODOLOGY

### Toluene permeation tube



### Gas generation system

Temperature of the permeation tube is regulated at  $\pm 0.1^\circ\text{C}$ . All the flow rates of dry zero-air are controlled with permeation oven with Mass-Flow Controllers (MFCs). MFC3 is used to suck a portion of the flow of MFC1 using a pump. The mixture is mixed using a mixing chamber and delivered to an analyzer (online TD-GC-FID) for analysis (Figure 3).

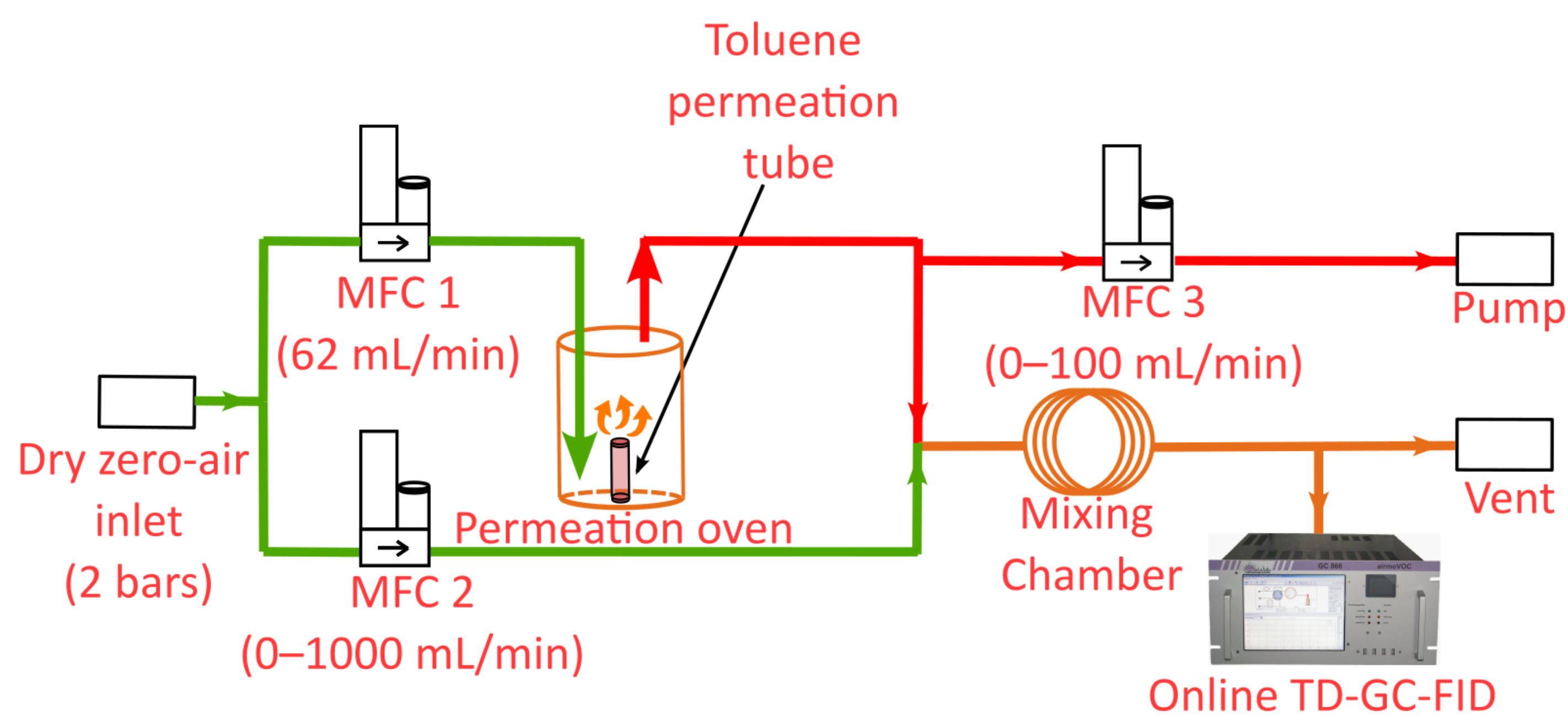


Figure 3: Generation setup of toluene emissions coupled to an online TD-GC-FID analyzer for chromatographic analysis.

### Online TD-GC-FID

Principle (Figure 4):

- preconcentration (adsorbent trap),
- separation (chromatographic column),
- and detection (FID).

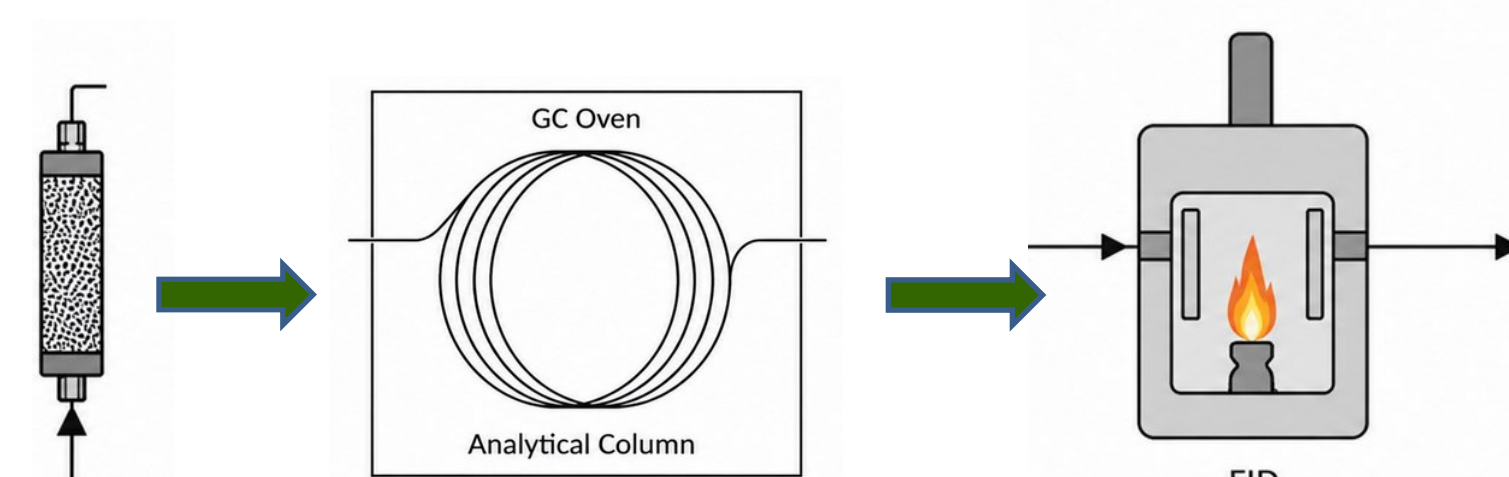


Figure 4: Setup of online TD-GC-FID.

## RESULTS AND DISCUSSION

### Permeation stability of toluene permeation tube

→ Continuous measurement of the concentration for  $T^\circ\text{-range} = 40\text{--}70^\circ\text{C}$  after 24-hr equilibration period in the permeation oven (Figure 5).

→ Evaluation of the emission rate  $ER(T)$  at  $40^\circ\text{C}$  based on Equation (1) after attaining stability.

$$ER(T = 40^\circ\text{C}) = 90.64 \pm 3.69 \text{ ng/min} \quad (RSD = 0.94\%)$$

$$\text{Equation (1): } C(\text{toluene}) = \frac{ER(T)}{F} \times \frac{22.4}{M} \times \frac{T}{273}$$

### Concentration range of emitted toluene

→ Concentration can be modified by varying  $T$ , MFC1 & MFC2 flow rates, and sucking MFC3 flow rate.

😊 Advantage of using the extra MFC3 (up to 84% portion of MFC1 flow is sucked out): obtaining a concentration down to 4.2 ppb.

### Temperature dependence of $ER(T)$

→ Two models to explain the temperature dependence (Figure 6):

1. Arrhenius-type model [4]

$$\ln ER(T) = -\frac{E_a}{R} \cdot \frac{1}{T} + \ln A$$

2. Antoine-Fickian model [5]

$$\ln ER(T) = A' + \frac{B}{T}$$

$$-\frac{E_a}{R} = -7270; \ln A = 27.92$$

$$B = -7270; A' = 27.92$$

### Accuracy of measurements on online TD-GC-FID

→ Measured and generated concentrations are linearly correlated ( $R^2 = 0.999$ , Figure 7).

→ Online TD-GC-FID is accurate and precise (repeatability < 2% RSD at 5 ppb).

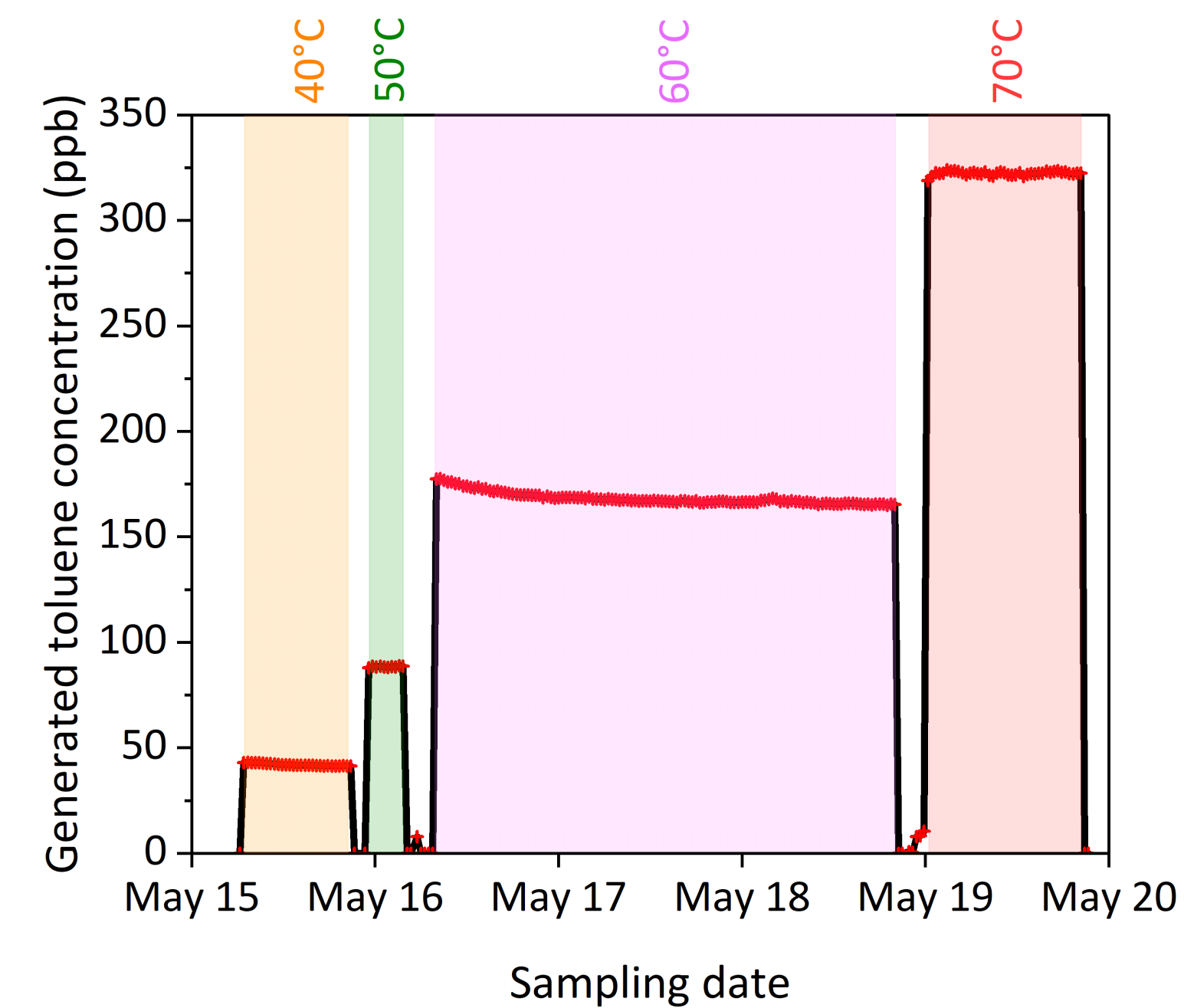


Figure 5: Generated concentrations of toluene (in ppb) at 4 different temperatures ( $40^\circ\text{C}$ ,  $50^\circ\text{C}$ ,  $60^\circ\text{C}$ , and  $70^\circ\text{C}$ ) for consecutive six days (total flow =  $624.4 \pm 12.66 \text{ mL/min}$ ).

→ 4.2 ppb (minimum at  $40^\circ\text{C}$ )  
4.1 ppm (maximum at  $70^\circ\text{C}$ )

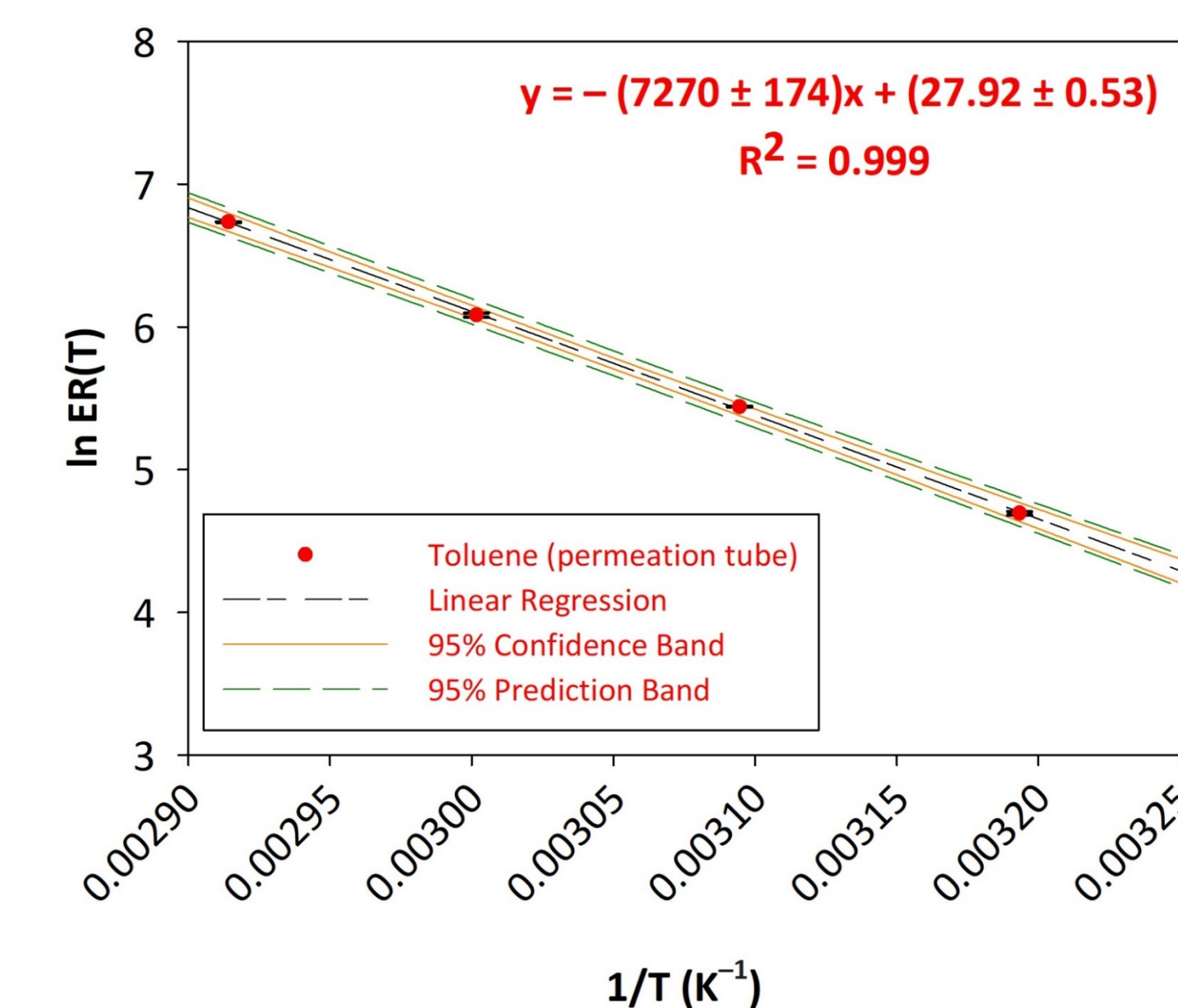


Figure 6: Temperature dependence of emission rates of toluene at 4 different temperatures ( $40^\circ\text{C}$ ,  $50^\circ\text{C}$ ,  $60^\circ\text{C}$ , and  $70^\circ\text{C}$ ) and its linear fit. Vertical error bars correspond to the uncertainty calculated from the errors of the three MFC flow rates.

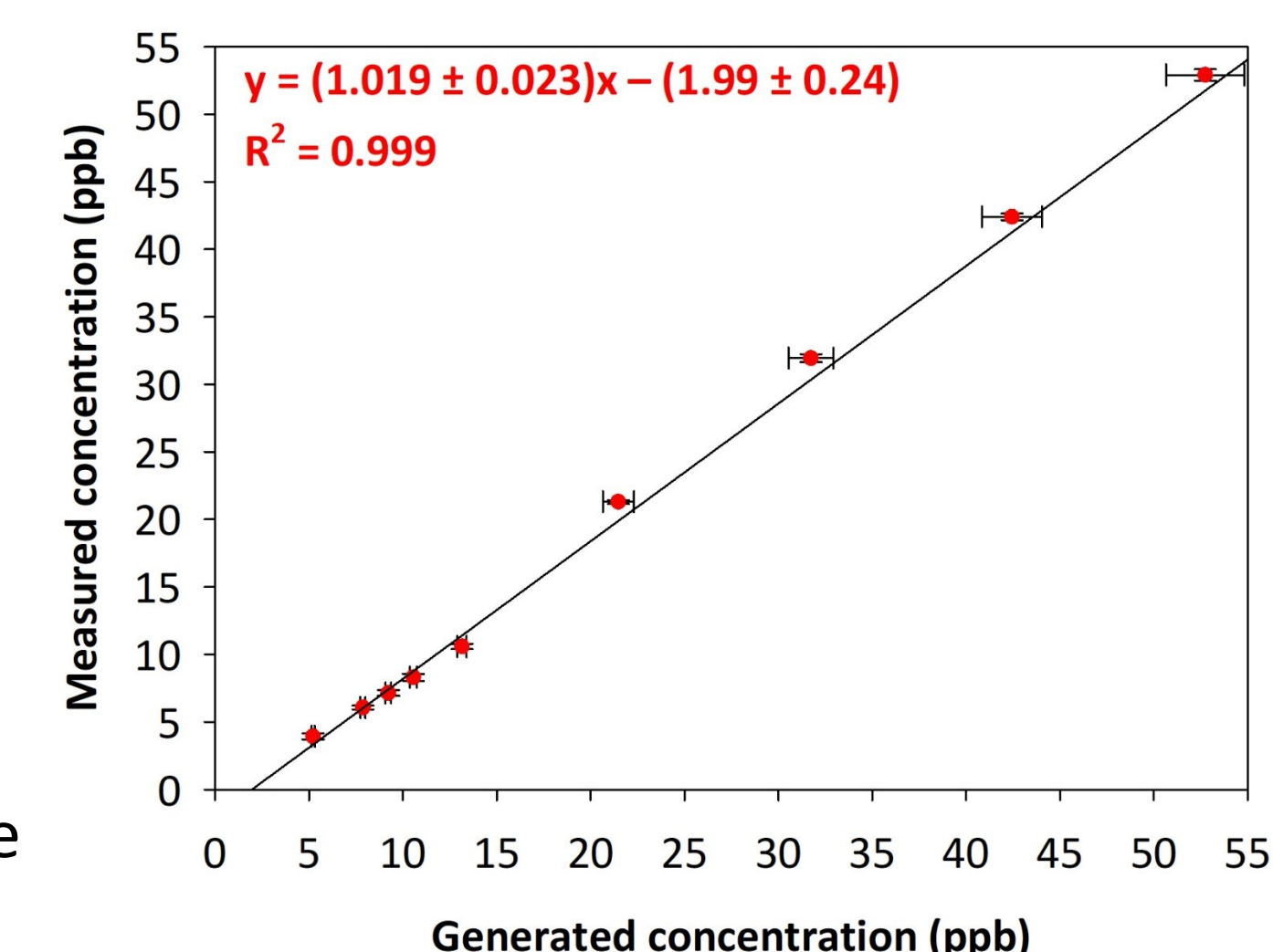


Figure 7: Measured concentration versus generated concentration of toluene at  $40^\circ\text{C}$ . Vertical error bars correspond to the standard deviation of  $n=10$  measurements in ppb.

## CONCLUSION

- Successful fabrication of toluene permeation tube and gas generation system.
- The temperature dependence was explained using two models.
- Concentrations from ppb to ppm levels (3 orders of magnitude) were attained by adjusting the oven temperature. In addition, MFC flow rates can also vary (dilution and sucking).
- Online TD-GC-FID has high accuracy in the detection of toluene generated concentration.
- Strong potential for extension to other BTEX and other VOCs that are not available in a pressurized cylinder.

## ACKNOWLEDGEMENTS

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[1] G. Li, S. Deng, J. Li, J. Gao, Z. Lu, X. Yi, J. Liu, Air Qual. Atmos. Health 15 (2022) 1941–1952.

[2] ATSDR, Toxicological profile for toluene, U.S. Dep. Health Hum. Serv., Atlanta (2000).

[3] N. Baimatova, B. Kenessov, J.A. Koziel, L. Carlsen, M. Bektassov, O.P. Demyanenko, Talanta 154 (2016) 46–52.

[4] R. Aoyagi, T. Arakawa, K. Iitani, K. Mitsubayashi, Y. Sekine, Y. Kaifuku, Discover Applied Sciences 7 (2025) 1–11.

[5] A. Grandjean, A. Becker, C. Kustner, M. Wolf, C. Sutter, R. Severac, J.P. Amiet, D. Bazin, S. Le Calvé, Microchemical Journal 206 (2024).